

## Appendix B.

### Conversion Factors

Conversions of concentrations of contaminants measured to comparison concentrations.

*General information for benzene, toluene, ethylbenzene and xylenes:*

1.0 milligram = 1,000 micrograms (1.0 mg = 1,000 µg)

1.0 µg = 0.001 mg

1,000 parts per billion (ppb) = 1.0 part per million (ppm)

1.0 ppb = 0.001 ppm

*Conversions from µg/m<sup>3</sup> to ppb:*

$$\text{ppb} = \frac{\text{concentration of chemical in } \mu\text{g/m}^3 \times 24.45}{\text{molecular weight of chemical}}$$

(24.45 is a constant, and is the volume in liters of one mole of gas.)

*Part 2. Specific conversions.*

Benzene:

Maximum concentration is 4.5 µg/m<sup>3</sup>

Molecular weight of benzene is 78.11

$$\text{ppb} = \frac{4.5 \mu\text{g/m}^3 \times 24.5}{78.11}$$

= 1.4 ppb benzene in air

Toluene:

Maximum concentration is 7.1 µg/m<sup>3</sup>

Molecular weight = 92.14

$$\text{ppb} = \frac{7.1 \mu\text{g/m}^3 \times 24.5}{92.14}$$

= 1.9 ppb toluene in air

Ethylbenzene:

Maximum concentration is 1.0 µg/m<sup>3</sup>

Molecular weight is 106.17

$$\text{ppb} = \frac{1.0 \mu\text{g}/\text{m}^3 \times 24.5}{106.17}$$

= 0.23 ppb ethylbenzene in air

Xylenes:

Maximum concentration is 2.8

Molecular weight is 106.16

$$\text{ppb} = \frac{2.8 \mu\text{g}/\text{m}^3 \times 24.5}{106.16}$$

= 0.65 ppb total xylenes in air