

4. CHEMICAL AND PHYSICAL INFORMATION

4.1 CHEMICAL IDENTITY

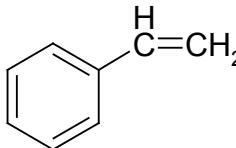
Information regarding the chemical identity of styrene is located in Table 4-1.

4.2 PHYSICAL AND CHEMICAL PROPERTIES

Information regarding the physical and chemical properties of styrene is located in Table 4-2.

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Table 4-1. Chemical Identity of Styrene

Characteristic	Information	Reference
Chemical name	Styrene	Verschueren 1983
Synonym(s)	cinnamene; cinnamol; ethenylbenzene; phenylethylene; styrol; vinylbenzene;	Verschueren 1983; HSDB 2009
Registered trade name(s)	No data	
Chemical formula	C ₈ H ₈	Windholz 1983
Chemical structure		IARC 1994
Identification numbers:		
CAS registry	100-42-5	Sax and Lewis 1987
NIOSH RTECS	WL3675000	HSDB 2009
EPA hazardous waste	No data	
EINICS	202-851-5	ESIS 2009
OHM/TADS	7216911	HSDB 2009
DOT/UN/NA/IMDG shipping	IMDG 3.3 UN 2055	HSDB 2009
HSDB	171	HSDB 2009
NCI	C02200	HSDB 2009

CAS = Chemical Abstracts Service; DOT/UN/NA/IMDG = Department of Transportation/United Nations/North America/International Maritime Dangerous Goods Code; EPA = Environmental Protection Agency; ESIS = European chemical Substances Information System HSDB = Hazardous Substances Data Bank; NCI = National Cancer Institute; NIOSH = National Institute for Occupational Safety and Health; OHM/TADS = Oil and Hazardous Materials/Technical Assistance Data System; RTECS = Registry of Toxic Effects of Chemical Substances

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Table 4-2. Physical and Chemical Properties of Styrene

Property	Information	Reference
Molecular weight	104.15	O'Neil et al. 2001 Lide 2005
Color	Colorless to yellowish	Windholz 1983
Physical state	Liquid	Sax and Lewis 1987
Melting point	-30.6 °C	O'Neil et al. 2001
Boiling point	145.2 °C	Verschueren 2001; Weast 1985
Density at 20 °C	0.9059	O'Neil et al. 2001
Odor	If pure, sweet and pleasant; commonly contains aldehydes which provide it with a penetrating, sharp, and unpleasant smell	Verschueren 2001
Odor threshold:		
Water	0.73 mg/L 0.011 mg/L	HSDB 2009 Amoore and Hautala 1983
Air	1.36 mg/m ³	Amoore and Hautala 1983
Solubility:		
Water at 15 °C	280 mg/L	Verschueren 2001
Water at 20 °C	300 mg/L	
Water at 40 °C	400 mg/L	
Organic solvents	Soluble in alcohol, ether, acetone, carbon disulfide	Windholz 1983
Partition coefficients:		
Log K _{ow}	2.95	Hansch et al. 1995; EPA 1984a
Log K _{oc}	2.96	Sabljic et al. 1995
Vapor pressure at 20 °C	5 mmHg	Verschueren 2001
Henry's law constant at 25 °C	2.61x10 ⁻³ atm-m ³ /mol (calculated)	EPA 1981
Autoignition temperature	914 °F (490 °C)	Sax and Lewis 1987
Flashpoint	87 °F (31 °C) (closed cup) 34.4°C (Tag open cup)	O'Neil et al. 2001; Kirk-Othmer 2001
Flammability limits	No data 0.9 (lower); 6.8 (higher) 1.1 (lower); 6.1 (higher)	CEFIC 2008; Kirk-Othmer 2001
Conversion factors	1 mg/m ³ =0.23 ppm; 1 ppm=4.33 mg/m ³	Verschueren 2001