## 4. CHEMICAL AND PHYSICAL INFORMATION

## 4.1 CHEMICAL IDENTITY

Information regarding the chemical identity of malathion is located in Table 4-1.

Malathion is manufactured in the United States as a technical-grade concentrate that is >90% pure malathion and contains approximately 5% of impurities consisting largely of reaction byproducts and degradation products. As many as 14 impurities have been identified in technical-grade malathion. The identities of the impurities and their percent (w/w) in technical grade malathion were found to be as follows: S-1,2-ethyl-*O*,*S*-dimethyl phosphorodithioate (isomalathion; 0.2%), S-1,2-bis(ethoxycarbonyl)-ethyl-*O*,*O*-dimethyl phosphorothioate (malaxon; 0.1%), diethylfumarate (DEF; 0.9%), *O*,*S*,*S*-trimethyl phosphorodithioate (0.003–1.2%), *O*,*O*,*S*-trimethyl phosphorothioate (0.04%), *O*,*O*,*S*-trimethyl phosphorodithioate (1.2%), *O*,*O*,*O*-trimethyl phosphorothioate (0.45%), diethylhydroxysuccinate (0.05%), ethyl nitrite (0.03%), diethyl mercaptosuccinate (0.15%), diethyl methylthiosuccinate (1.0%), *O*,*O*-dimethylphosphorothioate (0.05%), diethyl ethylthiosuccinate (0.1%), and sulfuric acid (0.05%). Malathion is formulated as an emulsifiable concentrate (EC), a dust (D), a wettable powder (WP), a ready-to-use (RTU) liquid, and a pressurized liquid. The quantity of active ingredient (ai) in EC and RTU formulations is variable and can contain up to 82 and 95%, respectively (Brown et al. 1993b; EPA 2001a).

## 4.2 PHYSICAL AND CHEMICAL PROPERTIES

Information regarding the physical and chemical properties of malathion is located in Table 4-2.

Table 4-1. Chemical Identity of Malathion

Characteristic	Information	Reference	
CAS Nomenclature	Diethyl[(dimethoxyphosphino -thioyl)thio]butanedioate	CAS 2001	
Common name	Malathion	Howard and Neal 1992	
Synonym(s)	1,2-Di(ethoxycarbonyl)ethyl O,O-dimethyl phosphorodithioate	Howard and Neal 1992	
Registered trade name(s)	Cekumal Fyfanon <sup>®</sup> Malixol <sup>®</sup> Maltox <sup>®</sup>	Farm Chemicals Handbook 2000 Howard and Neal 1992 Farm Chemicals Handbook 2000 Howard and Neal 1992	
Chemical formula	$C_{10}H_{19}O_6PS_2$	Howard and Neal 1992	
Chemical structure	Butanedioic acid, [(dimethoxyphosphinothioyl) thio]-, diethyl ester (malathion)		
Identification numbers: CAS registry NIOSH RTECS EPA hazardous waste OHM/TADS DOT/UN/NA/IMCO	000121-75-5 WM8400000 NA 2783; Malathion	Howard and Neal 1992 HSDB 2001 HSDB 2001	
shipping HSDB NCI	665	HSDB 2001	

CAS = Chemical Abstracts Services; DOT/UN/NA/IMCO = Department of Transportation/United Nations/North America/International Maritime Dangerous Goods Code; EPA = Environmental Protection Agency; HSDB = Hazardous Substances Data Bank; NCI = National Cancer Institute; NIOSH = National Institute for Occupational Safety and Health; OHM/TADS = Oil Hazardous Materials/Technical Assistance Data System; RTECS = Registry of Toxic Effects of Chemical Substance

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**Table 4-2. Physical and Chemical Properties of Malathion** 

Property	Information	Reference	
Molecular weight	330.36	Howard and Neal 1992	
Color	Colorless liquid (pure form)	Matsumura 1985	
	Deep brown to yellow	Budavari 1996; NIOSH 1997	
Physical state	Liquid	Matsumura 1985	
Melting point	2.9 °C	Budavari 1996	
Boiling point	156–157 °C	Budavari 1996	
Boiling point pressure	0.7 torr		
Density:	2		
at 25 °C	1.23 g/cm <sup>3</sup>	Budavari 1996	
Odor	Garlic-like	NIOSH 1997	
	Mercaptan	Farm Chemicals Handbook 1999	
Odor threshold:			
Water 60 °C	1.0 mg/L	Fazzalari 1978	
Air	13.5 mg/m <sup>3</sup> (low) and	Ruth 1986	
Calubility.	13.5 mg/m <sup>3</sup> (high)		
Solubility: Water at 20 °C	145 mg/l		
	145 mg/L	Tomlin 1997	
Organic solvent(s)	Miscible with alcohols, esters, ketones, ethers, aromatics,	Budavari 1996	
	and vegetable oil; limited	Budavan 1990	
	solubility in paraffin		
	hydrocarbons		
Partition coefficients:	,		
Log K <sub>ow</sub>	2.36	Hansch et al. 1995	
Log K <sub>ow</sub>	2.89	Chiou et al. 1977; Freed et al.	
Log K <sub>oc</sub>	3.25	1978	
-		Buyuksonmez 1999	
Vapor pressure			
at 25 °C	5.03x10 <sup>-6</sup> torr	Watanabe 1993	
at 30 °C	3.38x10 <sup>-6</sup> torr	SRC 2000	
at 25 °C	7.9x10 <sup>-6</sup> torr	Kim et al. 1984	
Henry's law constant (25	4.9x10 <sup>-9</sup> atm m <sup>3</sup> /mol	Fendinger et al. 1990	
°C)			
Autoignition temperature	No data		
Flashpoint	163 °C <sup>a</sup>	Farm Chemicals Handbook 1989	
Flammability limits	No data		
Conversion factors <sup>b</sup>	No data		
Explosive limits	Containers of malathion may	U.S. Coast Guard 1984–1985	
	explode in a fire		

<sup>&</sup>lt;sup>a</sup>Pensky-Martens closed cup test <sup>b</sup>The conversion factor for ppm to mg/m³ is: ppm = (mg/m³) (24.45 L/mole)/(g/mole).